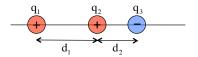
Problem 23.8

The charges and distances are shown, along with the Coulomb force calculations. Something to note before starting. Notice below that I



have not used -2.00×10^{-6} coulombs in my force calculations but, rather, the magnitude $+2.00 \times 10^{-6}$ coulombs. Why? Because positive and negative signs in force equations denote the DIRECTION of the force, and in Coulomb force equations that is determined by whether the two charges are attracted or repulsed by one another. Bottom line: Use Coulomb's Law to determine the magnitude of the Coulomb force using only the magnitude of the charges involved, then assign a sign relative to the coordinate axis, depending upon whether there is attraction or repulsion between the two charges.

With that in mind:

1.)

Our force quantities are:

$$\begin{array}{c|c} q_1 & q_2 & q_3 \\ \hline + & + & \hline \\ \hline d_1 & \times & d_2 \end{array}$$

$$F_{\text{on } q_2} = k \frac{q_1 q_2}{d_1^2} + k \frac{q_2 q_3}{d_2^2}$$

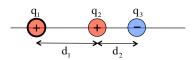
$$= k q_2 \left[\frac{q_1}{d_1^2} + \frac{q_3}{d_2^2} \right]$$

$$= (8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) (1.50 \times 10^{-6} \text{ C}) \left[\frac{(6.00 \times 10^{-6} \text{ C})}{(3.00 \times 10^{-2} \text{ m})^2} + \frac{(2.00 \times 10^{-6} \text{ C})}{(2.00 \times 10^{-2} \text{ m})^2} \right]$$

$$= 157 \text{ N}$$

3.)

Our force quantities are:



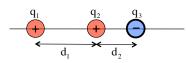
$$F_{\text{on }q_1} = -k \frac{q_1 q_2}{d_1^2} + k \frac{q_1 q_3}{(d_1 + d_2)^2}$$

$$= k q_1 \left[-\frac{q_2}{d_1^2} + \frac{q_3}{(d_1 + d_2)^2} \right]$$

$$= (8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) (6.00 \times 10^{-6} \text{ C}) \left[-\frac{(1.50 \times 10^{-6} \text{ C})}{(3.00 \times 10^{-2} \text{ m})^2} + \frac{(2.00 \times 10^{-6} \text{ C})}{(5.00 \times 10^{-2} \text{ m})^2} \right]$$

$$= -46.7 \text{ N}$$

Our force quantities are:



$$F_{\text{on } q_3} = -k \frac{q_3 q_2}{d_2^2} - k \frac{q_1 q_3}{\left(d_1 + d_2\right)^2}$$

$$= k q_3 \left[-\frac{q_2}{d_2^2} - \frac{q_3}{\left(d_1 + d_2\right)^2} \right]$$

$$= (8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) (2.00 \times 10^{-6} \text{ C}) \left[-\frac{\left(1.50 \times 10^{-6} \text{ C}\right)}{\left(2.00 \times 10^{-2} \text{ m}\right)^2} - \frac{\left(6.00 \times 10^{-6} \text{ C}\right)}{\left(5.00 \times 10^{-2} \text{ m}\right)^2} \right]$$

$$= 111 \text{ N}$$

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